KONGUNADU ARTS AND SCIENCE COLLEGE (AUTONOMOUS)

Re-accredited by NACC with 'A+' Grade (4th Cycle)

College of Excellence (UGC)

GN Mills Post, Coimbatore – 641 029



DEPARTMENT OF CHEMISTRY (UG)

COURSE OUTCOMES (CO) OF B.Sc., CHEMISTRY

CURRICULUM AND SCHEME OF EXAMINATIONS (CBCS)(2023 – 2024 and onwards)

COURSE OUTCOMES (CO)

| Programme Code: 04 | | B.Sc. Chemistry | | | | |
|--------------------|-----------|---|--|---------------------------|--------------|--|
| Tit | le of the | paper | CORE PAPER – I: INORGANIC, ORGANIC AND PHYSICAL CHEMISTRY – I | | | |
| Batch | 1 | Semester | Hours / Week | Total Hours | Credits | |
| 2023 - 20 | 024 | I | 6 | 90 | 5 | |
| | | | Objectives | | | |
| 1. | To kno | ow the concept | of qualitative inorganic ar | nalysis. | | |
| 2. | To acq | acquaint knowledge about electron displacement effects, hybridization and | | | | |
| 2. | conform | mations. | | | | |
| 3. | To kno | w about the str | ructure of an atom. | | | |
| | | | | | | |
| | | | Course Outcomes (| CO) | | |
| | CO1 | Explain the b | asic analytical knowledge | e and group separation of | of elements. | |
| | CO2 | To know the | types of bonding and geor | metry in molecules and | VSEPR theory | |
| K1 – K5 | CO3 | Explain the is | somerism of alkanes and o | cycloalkanes. | | |
| | CO4 | Acquire the k | nowledge about the struc | ture of atoms. | | |
| | CO5 | Understand c | Understand characteristics of gases. | | | |

Subject code: 23EVS101

| Programme code: 04 | | | | | | |
|---------------------------------|-------------|--|--|--|--|--|
| PART IV – ENVIRONMENTAL STUDIES | | | | | | |
| Hours / Week | Total Hours | | | | | |
| 2 | 30 | | | | | |

Course Objectives

The course will provide students with an understanding and appreciation of the complex interactions of man, health and the environment. It will expose students to the multi-disciplinary nature of environmental health sciences

To inculcate knowledge and create awareness about ecological and environmental concepts, issues and solutions to environmental problems.

To shape students into good "Ecocitizens" thereby catering to global environmental needs.

This course is designed to study about the types of pollutants including gases, chemicals petroleum, noise, light, global warming and radiation as well as pollutant flow and recycling and principles of environmental pollution such as air, water and soil

The course will address environmental stress and pollution, their sources in natural and workplace environments, their modes of transport and transformation, their ecological and public health effects, and existing methods for environmental disease prevention and remediation

Course Outcomes (CO)

Understand how interactions between organisms and their environments drive the dynamics of individuals, populations, communities and ecosystems.

Develop an in depth knowledge on the interdisciplinary relationship of cultural, ethical and social aspects of global environmental issues.

Acquiring values and attitudes towards complex environmental socio-economic challenges and providing participatory role in solving current environmental problems and preventing the future ones.

To gain inherent knowledge on basic concepts of biodiversity in an ecological context and about the current threats of biodiversity.

To appraise the major concepts and terminology in the field of environmental pollutants, its interconnections and direct damage to the wildlife, in addition to human communities and ecosystems.

| | | | | Subject c | oue: 250CH202 | |
|--------------------|-----------|----------------|----------------------------|--------------------------|---------------|--|
| Programme Code: 04 | | | | B.Sc. Chemistry | | |
| Tit | le of the | paper | CORE PAPER – II | | | |
| | | | INORGANIC, ORGAN | NIC AND PHYSICAL C | CHEMISTRY -II | |
| Batch | 1 | Semester | Hours / Week | Total Hours | Credits | |
| 2023 - 20 | 024 | II | 6 | 90 | 5 | |
| | l | | 1 | | | |
| | | | Course Objective | \mathbf{s} | | |
| 1 | Know | about metallur | gy, importance of periodic | table and atomic prope | erties. | |
| 2 | To lear | n about Benze | ene and Aromaticity. | | | |
| 3 | To stud | dy the fundame | entals of thermodynamics | and thermochemistry. | | |
| | | | | | | |
| | | | Course Outcomes (C | CO) | | |
| | GO1 | Obtain probl | em solving skills in order | to modify industrial pro | ocesses in | |
| | CO1 | Extraction m | etallurgy. | | | |
| 17.1 17.5 | CO2 | Gain knowle | dge about periodic propert | ties | | |
| K1 – K5 | CO3 | Study of Aro | omatic Compounds and me | echanism of certain reac | ctions | |
| | CO4 | Learn about | concepts of thermodynami | ics. | | |
| | CO5 | Acquire the l | knowledge in thermochem | istry. | | |

Subject code: 23VED201

| | | | | | ode: 23VED201 |
|-------------------------------|-----------|-------------------|------------------------------|--------------------------|---------------------|
| Progr | ramme (| Code: 04 | B.Sc. Chemistry | | |
| Tit | le of the | paper | MORAL AND ETHICS | | |
| Batch | 1 | Semester | Hours / Week | Total Hours | Credits |
| 2023 - 20 | 024 | II | 2 | 30 | 2 |
| | | | | | |
| | | | Course Objectives | S | |
| 1. | To imp | oart Value Edu | cation in every walk of life | 2. | |
| 2. | To help | p the students t | o reach excellence and rea | p success. | |
| 3. | To imp | part the right at | titude by practicing self-in | trospection. | |
| 4. | To por | tray the life an | d messages of Great Leade | ers. | |
| 5. | To insi | ist the need for | universal brotherhood, par | tience and tolerance. | |
| 6. | To help | p the students t | o keep them fit. | | |
| 7. | To edu | cate the impor | tance of Yoga and Meditat | tion. | |
| | | | | | |
| | | | Course Outcomes (C | CO) | |
| | CO1 | Will be able | to recognize Moral values, | Ethics, contribution of | leaders, Yoga and |
| | COI | its practice | | | |
| | CO2 | Will be able | to differentiate and relate | the day to day applicat | tions of Yoga and |
| | CO2 | Ethics in real | life situations | | |
| K1 – K5 | CO3 | Can emulate | the principled life of great | warriors and take it for | ward as a message |
| $\mathbf{K}_1 - \mathbf{K}_2$ | CO3 | to self and th | e society | | |
| | CO4 | Will be able | to Analyse the Practical or | atcome of practicing M | oral values in real |
| | CO4 | life situation | | | |
| | CO5 | Could Evalu | ate and Rank the outcom | ne of the pragmatic ap | proach to further |
| | 003 | develop the s | kills | | |

| | | | | Subject co | de: 23UCH2CL |
|--------------------|--------------------|-----------------|---|---------------------------|--------------|
| Programme Code: 04 | | | B.Sc. Chemistry | | |
| Tit | Title of the paper | | CORE PRACTICAL – I | | |
| | | | INORGANIC Q | UALITATIVE ANAL | YSIS AND |
| | | |] | PREPARATIONS | |
| Batch | ı | Semester | Hours / Week | Total Hours | Credits |
| 2023 - 2 | 024 | II | 3 | 90 | 2 |
| | • | | | | |
| | | | Course Objectiv | es | |
| 1. | To de | monstrate the b | asic laboratory techniqu | e of semi micro qualitati | ve analysis. |
| 2. | To un | derstand about | the interfering anions, it | s elimination and group | separation. |
| 3. | To pre | epare inorganic | complexes. | | |
| | 1 | | | | |
| | | | Course Outcomes (| (CO) | |
| | CO1 | Build the kn | owledge in principles of | semi micro qualitative a | nalysis. |
| | CO2 | Know about | the interfering and non- | interfering anions. | |
| V1 V5 | CO2 | Experience t | to remove interfering anion and group separation of various | | |
| K1 – K5 | CO3 | cations. | | | |
| | CO4 | Group separa | ation of various cations | | |
| | CO5 | Learn the pre | eparation of inorganic co | omplexes. | |

| | | | T | · · · · · · · · · · · · · · · · · · · | ode: 23UCH303 |
|--------------------|--------------------|--------------------------|--|---------------------------------------|-----------------|
| Programme Code: 04 | | | B.Sc. Chemistry | | |
| Tit | Title of the paper | | CO INORGANIC, ORGAN | ORE PAPER – III IC AND PHYSICAL CI | HEMISTRY - III |
| Batch | 1 | Semester | Hours / Week | Total Hours | Credits |
| 2023 - 20 | 024 | III | 4 | 60 | 4 |
| | | | Course Objective | es | |
| 1. | To kno | w the basic co | oncepts in quantitative anal | lysis. | |
| 2. | To obs | | istry of dicarboxylic acids | and reactions involving | g carbonyl |
| 3. | To enu equilib | | law of thermodynamics, s | state functions S, A, G a | and chemical |
| | | | Course Outcomes (Course Outcomes) | CO) | |
| | CO1 | Gain knowle volumetric a | dge in preparation, standa nalysis. | rdization of solution and | d principles of |
| K1 – K5 | CO2 | 1 - | Study the preparation, properties and reactions of di carboxylic acids, unsaturated acids and hydroxy acids. | | |
| | CO3 | To Study on | the preparation and proper | rties of aldehydes and k | tetones. |
| | CO4 | Analyze and | apply laws of thermodyna | amics. | |
| | CO5 | To understar | nd the importance of absolu | ute zero | |

Subject code: 23UGC3S1

| _ | | | | Subject co | oae: 23UGC381 | | | |
|-----------|-------------------|--|--|-------------------------|-------------------|--|--|--|
| Prog | ramme (| Code: 04 | B.Sc. Chemistry | | | | | |
| Tit | le of the | paper | Skill Based Subject 1 – Cyber Security | | | | | |
| Batch | 1 | Semester | Hours / Week | Total Hours | Credits | | | |
| 2023 - 20 | 024 | III | 2 | 30 | 3 | | | |
| | Course Objectives | | | | | | | |
| 1. | The co | urse introduces | the basic concepts of Cyl | ber Security | | | | |
| 2. | To dev | elop an ability to understand about various modes of Cyber Crimes and Preventive | | | | | | |
| 2. | measur | res | | | | | | |
| 3. | To und | lerstand about t | he Cyber Legal laws and | Punishments | | | | |
| | | | | | | | | |
| | | | Course Outcomes (C | C O) | | | | |
| | CO1 | To Understan | d the Concepts of Cyberc | rime and Cyber Frauds | | | | |
| 17.1 17.5 | CO2 | To Know abo | out Cyber Terrorism and i | its preventive measures | | | | |
| K1 – K5 | CO3 | To Analyze a | bout the Internet, Mobile | Phone and E-commerce | e security issues | | | |
| | CO4 | To Understan | d about E-mail and Socia | l Media Issues | | | | |
| | CO5 | To Describe a | about various legal respon | ses to Cybercrime | | | | |

| | | | | Subject C | oue: 250CH404 |
|--------------------|-----------|--|------------------------------|--------------------------|------------------|
| Programme Code: 04 | | | B.Sc. Chemistry | | |
| Tit | le of the | nonor | CORE PAPER – IV | | |
| 110 | ie oi uie | paper | INORGANIC, ORGAN | IC AND PHYSICAL C | HEMISTRY - IV |
| Batch | ı | Semester | Hours / Week | Total Hours | Credits |
| 2023 - 20 | 024 | IV | 4 | 60 | 4 |
| | • | | | , | |
| | | | Course Objective | S | |
| 1. | To lear | n group IA ele | ements. | | |
| 2. | To kno | w about vario | us types of alcohols, pheno | ols and their reactions | |
| 3. | To kno | w about phase | rule and phase equilibria | | |
| | | | | | |
| | | | Course Outcomes (| C O) | |
| | CO1 | Gain the kno | wledge about the propertion | es of alkali metals. | |
| | CO2 | Understand the basic aspects of phenols, amines and its derivatives. | | | |
| 17.1 17.5 | CO3 | Analyze and | apply phase rule to variou | is systems. | |
| K1 – K5 | CO4 | Understand of | colligative properties and t | heir determinations. | |
| | G0. | Understandin | ng Intellectual Properties a | and the importance of it | and awareness of |
| | CO5 | patents | | | |

| Subject code, 250 C11452 | | | | | |
|--------------------------|--------------------|---|--------------------------------|------------------|---------|
| Programme Code: 04 | | | B.Sc. Chemistry | | |
| T:4 | Title of the paper | | SKILL BASED SUBJECT-II | | |
| 110 | ie oi tiie | paper | WATER POLLUTION AND MANAGEMENT | | |
| Batch | 1 | Semester | Hours / Week | Total Hours | Credits |
| 2023 - 20 | 024 | IV | 2 | 30 | 3 |
| | | | | | |
| | | | Course Objective | es | |
| 1. | To kno | ow about the s | ources and characteristics | of water. | |
| 2. | To lear | n about the ana | alysis of the pollutants in | water. | |
| 3. | To lear | n the methods | of purification and manag | gement of water. | |
| | | | | | |
| | | | Course Outcomes (| C O) | |
| | CO1 | To understan | d the importance of water | : | |
| | CO2 | To study the types of water pollution. | | | |
| K1 – K5 | CO3 | To analyze and measure the toxic chemical substances. | | | |
| | CO4 | To gain know | ledge in purification tech | niques of water. | |
| | CO5 | To know the irrigation systems used in agriculture. | | | |

| Subject code. 250 cm²-cm² | | | | | uc. 250 C11+C1/1 | | | |
|---------------------------|-----------|--------------------|----------------------------------|------------------------|--|--|--|--|
| Programme Code: 04 | | | B.Sc. Chemistry | | | | | |
| | | | CORE PRACTICAL – II | | | | | |
| Tit | le of the | paper | INORGANIC VOLUMETRIC AND ORGANIC | | | | | |
| | | | QUAL | ITATIVE ANALYSIS | 8 | | | |
| Batch | ı | Semester | Hours / Week | Total Hours | Credits | | | |
| 2023 - 20 | 024 | IV | 3 | 90 | 3 | | | |
| | , | 1 | | - | | | | |
| | | | Course Objective | S | | | | |
| 1. | To den | nonstrate the co | ncept of quantitative volu | ımetric analysis. | | | | |
| 2. | To unc | derstand the var | ious types of titrimetric ar | nalysis. | | | | |
| 3. | To ide: | ntify the function | onal groups of unknown o | rganic compounds. | | | | |
| | • | | | | | | | |
| | | | Course Outcomes (C | CO) | | | | |
| | CO1 | Gain the know | vledge in principles of vo | lumetric analysis. | | | | |
| | CO2 | Estimating the | e amount of substances pr | resent in solutions. | | | | |
| K1 – K5 | CO3 | Learn to appr | oach a problem systemati | cally | | | | |
| | CO4 | Interpret the r | esults logically. | | | | | |
| | CO5 | Detect variou | s functional groups preser | nt in an organic compo | Detect various functional groups present in an organic compound. | | | |

| Programme Code: 04 | | B.Sc. Chemistry | | | | |
|-------------------------------|------------|--|----------------------------|---|---------------------|--|
| | | | CORE PAPER – V | | | |
| Tit | tle of the | paper | SPECTROSCOP | Y AND CHROMATO | GRAPHIC | |
| | | | | TECHNIQUES | | |
| Batch | 1 | Semester | Hours / Week | Total Hours | Credits | |
| 2023 - 20 | 024 | V | 3 | 45 | 3 | |
| | | | Course Objective | ac. | | |
| | To kno | yw about the rec | | | of ultra violet | |
| 1. | | _ | nd Infrared spectroscopy | n of electromagnetic spectrum, fundamentals of ultra – violet | | |
| 2. | | | netic Resonance (NMR) | | 1 | |
| 3. | | · | terpret and solve problem | * ** | | |
| 3. | Specific | ometry and to n | nerpret and sorve problem | ns using various speed | u. | |
| | | | Course Outcomes (| CO) | | |
| | CO1 | Understand th | e basic principles of UV | -Visible spectroscopy a | nd to utilize their | |
| | COI | basic aspects to identify various organic compounds. | | | | |
| | CO2 | Gain the know | vledge in principles, and | functions of IR spectros | scopy. | |
| | CO3 | Study the basi | c principles of NMR spe | ctroscopy and apply to | identify the | |
| K1 – K5 | COS | organic comp | ounds. | | | |
| $\mathbf{K}_1 - \mathbf{K}_2$ | | Know about b | asic principles of mass s | pectroscopy technique a | and the | |
| | CO4 | application of | various spectral technique | ues to elucidate the stru | cture of organic | |
| | | molecules. | | | | |
| | CO5 | Exploring the | various chromatography | techniques and their ap | oplications in | |
| | 003 | separation of organic mixtures. | | | | |

| | | | | Subject co | ode: 23UCH506 | |
|-------------------------------|-----------|---|---|---|----------------|--|
| Programme Code: 04 | | | B.Sc. Chemistry | | | |
| Tr: | mid ca | | CORE PAPER – VI | | | |
| 110 | le of the | paper | INORGANIC CHEMISTRY | | | |
| Batch | ı | Semester | Hours / Week | Total Hours | Credits | |
| 2023 - 2 | 024 | V | 3 | 45 | 3 | |
| | • | | | | | |
| | | | Course Objective | es | | |
| 1. | To und | lerstand the key | features of coordination compounds, including: the variety of | | | |
| 1. | structu | res, ligands, va | rious theories of coordina | coordination complexes, stability of complexes. | | |
| 2. | To ide | ntify what radio | oisotopes and acquaint kn | owledge about nuclear i | reactions. | |
| 3. | To des | cribe about Ino | rganic acids, bases, Inorg | anic Solvents and Inorg | anic Polymers. | |
| | | | | | | |
| | | | Course Outcomes (Course Outcomes) | C O) | | |
| | CO1 | Understand th | ne theories of co-ordination | on compounds. | | |
| | CO2 | Knowledge a | bout basics nuclear Chen | nistry | | |
| K1 – K5 | CO3 | Analyze the i | mportance of radioactive | isotopes and nuclear rea | actions. | |
| $\mathbf{K}_1 - \mathbf{K}_2$ | CO4 | Describe abou | at the different concepts of | of Inorganic acids, bases | , Inorganic | |
| | 004 | Solvents and | Inorganic Polymers. | | | |
| | CO5 | Understanding the importance of Inorganic Solvents and Inorganic Poly | | | | |

| | Subject code. 250CH507 | | | | | | |
|-----------|------------------------|-----------------|---|---------------------------|-------------------|--|--|
| Prog | ramme (| Code: 04 | | B.Sc. Chemistry | | | |
| TD: 4 | 1 6.1 | | CORE PAPER – VII | | | | |
| Tit | Title of the paper | | ORGANIC REACTION MECHANISM | | | | |
| Batch | ı | Semester | Hours / Week | Total Hours | Credits | | |
| 2023 - 20 | 024 | V | 4 | 60 | 3 | | |
| | | | Course Objective | S | | | |
| | | | | | | | |
| 1. | To stu | dy asymmetry a | and optical activity of or | ganic molecules and ba | asics in | | |
| 1. | carboh | ıydrate. | | | | | |
| 2. | To understand the m | | echanisms of important organic rearrangements reactions and | | | | |
| ۷. | Prepar | ations and reac | ions of Amines and Diaz | o compounds. | | | |
| 3. | To stu | dy preparation | and properties of heterocy | clic compounds. | | | |
| | | | Course Outcomes (C | CO) | | | |
| | | | | | | | |
| | CO1 | Understandin | g the fundamental aspects | s of stereochemistry. | | | |
| | CO2 | Learn about p | reparation, properties and | d structural elucidation | of carbohydrates. | | |
| *** | CO3 | Study on the | various naming reactions | and their detailed mech | hanistic pathway. | | |
| K1 – K5 | CO4 | | nowledge about the prepa | arations and reactions of | of Amines and | | |
| | | Diazo compo | | | | | |
| | CO5 | | knowledge about five and | six membered hetero | ocyclic | | |
| | | compounds | | | | | |

| | | | | | oue: 250Cn508 | | |
|-----------|--------------------|-------------------|----------------------------|---------------------------|----------------|--|--|
| Prog | ramme (| Code: 04 | | B.Sc. Chemistry | | | |
| Tr' | 1 6.1 | | CORE PAPER - VIII | | | | |
| 111 | Title of the paper | | PHYSICAL CHEMISTRY - I | | | | |
| Batch | Batch Semester | | Hours / Week | Total Hours | Credits | | |
| 2023 - 20 | 024 | V | 4 | 60 | 4 | | |
| | | | | | | | |
| | | | Course Objective | es | | | |
| 1. | To unc | lerstand the fund | lamentals of electrochem | nistry. | | | |
| 2. | To kno | ow the types and | importance of electrode | s and electro chemical of | cells. | | |
| 3. | To stud | dy about corrosi | on, batteries and Electro | analysis. | | | |
| | | | | | | | |
| | | | Course Outcomes (| C O) | | | |
| | CO1 | Understanding | the concept of conducta | ance and its applications | S. | | |
| | CO2 | Acquire basic | knowledge about electro | de potential, electroche | mical cell and | | |
| | CO2 | potentiometric | titrations. | | | | |
| K1 – K5 | CO3 | Understanding | the fundamental princip | oles of corrosion, protec | tive coatings | | |
| | CO4 | electroplating | and its significance. | | | | |
| | CO5 | Know about b | asic principles and instru | mentation of Electroche | emical Power | | |
| | | Systems, Polar | rography and its applicat | ions. | | | |

| | | | | Subject c | code: 23UCH609 | | |
|-----------|--------------------|--|--|--------------------------|------------------|--|--|
| Prog | ramme (| Code: 04 | | B.Sc. Chemistry | | | |
| Tr: | 1 C 41 | | CO | ORE PAPER – IX | | | |
| 110 | Title of the paper | | SOLID STATE ANI | D COORDINATION | CHEMISTRY | | |
| Batch | ı | Semester | Hours / Week | Total Hours | Credits | | |
| 2023 - 20 | 024 | VI | 5 | 75 | 4 | | |
| | • | | | | | | |
| | | | Course Objective | S | | | |
| 1. | To kno | w about funda | mentals of crystallography | and solid state Chemi | istry | | |
| 2. | To stud | To study about reactions of complexes. | | | | | |
| 3. | To in s | in sight knowledge about Bio – Inorganic Chemistry | | | | | |
| | 1 | | | | | | |
| | | | Course Outcomes (C | CO) | | | |
| | CO1 | Knowing the | difference between amorg | phous and crystalline so | olids and their | | |
| | CO1 | arrangement | in crystal lattice. | | | | |
| | CO2 | Learn about | defects in crystals, various | theories of metallic bo | onding and | | |
| 17.1 17.5 | CO2 | alloys. | | | | | |
| K1 – K5 | CO2 | Decide the v | arious crystal structures us | ing X-ray diffraction to | echniques and | | |
| | CO3 | Study about liquid crystals. | | | | | |
| | CO4 | Study about | various ligand substitution | reactions. | | | |
| | CO5 | To acquire k | To acquire knowledge about bioinorganic chemistry. | | | | |

| | | Т | | Subject co | ode: 23UCH010 | | |
|------------|--------------------|------------------|--|---------------------------|---------------|--|--|
| Prog | ramme (| Code: 04 | B.Sc. Chemistry | | | | |
| TP' 4 | 1 6.1 | | CORE PAPER – X | | | | |
| 110 | Title of the paper | | CHEMISTRY | OF NATURAL PRO | DUCTS | | |
| Batch | ı | Semester | Hours / Week | Total Hours | Credits | | |
| 2023 - 20 | 024 | VI | 5 | 75 | 4 | | |
| | • | <u>.</u> | Course Objective | es | | | |
| | | | | | | | |
| 1. | To stud | dy about Terper | noids and Alkaloids. | | | | |
| 2. | To und | lerstand about | Vitamins and Hormones. | | | | |
| 3. | To stud | dy the preparati | ions and reactions of amines, Diazocompounds and | | | | |
| <i>J</i> . | Chemo | otheraphy. | | | | | |
| | | | Course Outcomes (| C O) | | | |
| | | | | | | | |
| | CO1 | Study on the | classification, structural e | lucidation and synthesis | s of few | | |
| | | important ter | | | | | |
| V1 V5 | CO2 | Learn about s | tructural determination ar | nd synthesis of alkaloids | S. | | |
| K1 – K5 | CO3 | Acquire basic | knowledge about vitami | ns and hormones. | | | |
| | CO4 | To study about | ut Amino acids, peptides | and Proteins. | | | |
| | CO5 | To gain know | vledge about chemotherapy. | | | | |

| | | | | Bubjecte | 0ue. 230 C11011 | | | |
|-----------|------------|--|---|--------------------------|------------------|--|--|--|
| Prog | ramme (| Code: 04 | B.Sc. Chemistry | | | | | |
| Tr' | 1 C.1 | | CORE PAPER – XI | | | | | |
| Tit | tle of the | paper | PHYSICAL CHEMISTRY - II | | | | | |
| Batch | ı | Semester | Hours / Week | Total Hours | Credits | | | |
| 2023 - 20 | 024 | VI | 4 | 60 | 4 | | | |
| | • | | | | | | | |
| | | | Course Objective | S | | | | |
| 1. | To uno | To understand the basics and theoretical aspects of Chemical kinetics. | | | | | | |
| 2. | To lea | rn about kinetio | es of thermal and photoche | emical reactions. | | | | |
| 3. | To gai | n knowledge al | bout importance of catalys | is, colloids and Liquid | state. | | | |
| | | | | | | | | |
| | | | Course Outcomes (Course Outcomes) | CO) | | | | |
| | CO1 | Understand t | he basic principles, variou | s experimental technique | ues and | | | |
| | COI | Theories of o | hemical kinetics. | | | | | |
| | CO2 | To understan | d the importance of variou | as theories explaining c | hemical kinetic. | | | |
| K1 – K5 | CO3 | Gain the kno | wledge about principles of | f photochemical and ph | otosensitized | | | |
| | Process. | | | | | | | |
| | CO4 | Study the bas | sic principles and types of | catalysis and colloids. | | | | |
| | CO5 | Explore the f | Explore the fundamentals of Liquid State. | | | | | |

| | | | | | oue: 230 CH083 | | |
|-----------|--------------|---|-----------------------------|---------------------------|------------------|--|--|
| Prog | ramme (| Code: 04 | | B.Sc. Chemistry | | | |
| Tit | tle of the | e paper | SKILL BASED SU | BJECT – III FOOD C | CHEMISTRY | | |
| Batch | 1 | Semester | Hours / Week | Total Hours | Credits | | |
| 2023 - 20 | 23 - 2024 VI | | 2 | 30 | 3 | | |
| | | | | | | | |
| | | | Course Objective | S | | | |
| 1. | To hav | ve an idea about | food adulteration and foo | od preservation techniq | ues. | | |
| 2. | To und | o understand the chemistry of vinegar, fruit juices, vegetable acids and beverages. | | | | | |
| 3. | To ana | alyse and charac | eterize chemical aspects of | f milk. | | | |
| | | | | | | | |
| | | | Course Outcomes (C | CO) | | | |
| | CO1 | Know about t | he nutrition values in food | d, food adulteration, sta | andards of food, | | |
| | COI | contamination | n and food poisoning. | | | | |
| 17.1 17.5 | CO2 | Understand a | bout the minerals in food | | | | |
| K1 – K5 | CO3 | Know about f | Food additives | | | | |
| | CO4 | Understand th | ne detailed information ab | out commercially impo | ortant beverages | | |
| | CO5 | Know about o | dairy products | | | | |

| | | | | Subject co | ie: 230CH0CN | | |
|-----------|--------------------|------------------|--|---------------------------|--------------|--|--|
| Prog | Programme Code: 04 | | | B.Sc. Chemistry | | | |
| Tr: | Title of the paper | | CORE PRACTICAL – III | | | | |
| 111 | | | INORGANIC QUANTITATIVE ANALYSIS | | | | |
| Batch | ı | Semester | Hours / Week | Total Hours | Credits | | |
| 2023 - 20 | 024 | VI | 3 | 90 | 3 | | |
| | • | | | | | | |
| | | | Course Objective | es s | | | |
| 1. | To uno | derstand the con | cept of gravimetric analy | sis. | | | |
| 2. | To get | acquainted with | the experimental proced | lure of gravimetric analy | ysis. | | |
| 3. | To det | ermine the quan | tity of analyte in solution | 1. | | | |
| | | | | | | | |
| | | | Course Outcomes (Course Outcomes) | C O) | | | |
| | CO1 | Understand th | e basic principles of Grav | vimetric analysis. | | | |
| | CO2 | Understand al | out the various precipita | ting agents. | | | |
| K1 – K5 | CO3 | Determination | of analyte masses through | gh the gravimetric analy | vsis. | | |
| | CO4 | Improve the a | ccuracy of analysis. | | | | |
| | CO5 | To gain know | o gain knowledge about Metal analysis in cosmetic products using AAS | | | | |

| | | | | Subject co | de: 23UCH6CO | |
|-----------------------------|--|--------------------------------------|------------------------------------|-------------------------|------------------|--|
| Prog | ramme (| Code: 04 | | B.Sc. Chemistry | | |
| Ti | 10 of 41. | | CORE PRACTICAL – IV | | | |
| 110 | Title of the paper | | PHYSI | CAL EXPERIMENT | S | |
| Batch | Batch Semester | | Hours / Week | Total Hours | Credits | |
| 2023 - 2 | 024 | VI | 3 | 90 | 3 | |
| | ' | | | | | |
| | | | Course Objective | S | | |
| 1. | Transf | ormation of the | oretical knowledge gain to | o practical aspects. | | |
| 2. | To hav | ve experience ir | handling electrical and n | on-electrical equipmen | ts. | |
| 3. | To determine the strength of various solutions through spectrometric and | | | | | |
| | electro | chemical techn | iques. | | | |
| | | | | ~ ~ ` | | |
| | | | Course Outcomes (C | CO) | | |
| | CO1 | The results of | f physical chemistry exper | riments are incorporate | d in both | |
| | COI | theoretical an | d practical aspects. | | | |
| | CO2 | Gain familian | ity with a variety of physi | co-chemical measurem | nent techniques. | |
| K1 – K5 | CO3 | Interpret data | from an experiment, incl | uding the construction | of appropriate | |
| $\mathbf{K}1 - \mathbf{K}3$ | CO3 | graphs and the evaluation of errors. | | | | |
| | CO4 | To know abo | ut Determination of Cell (| Constant, Specific cond | luctivity and | |
| | CO4 | Equivalent co | onductivity of strong electrolyte. | | | |
| | CO5 | To determine | strength of acids and base | es by Conductometric | Γitration. | |

| | | | | Subject co | ue: 230CH0CF | | |
|----------|--|---|---------------------------|---------------------------|--------------|--|--|
| Prog | ramme (| Code: 04 | | B.Sc. Chemistry | | | |
| T: | lo of the | nonor | CORE PRACTICAL – V | | | | |
| 110 | Title of the paper | | APPLICATIO | ON ORIENTED PRAC | TICAL | | |
| Batch | 1 | Semester | Hours / Week | Total Hours | Credits | | |
| 2023 - 2 | 024 | VI | 4 | 120 | 4 | | |
| | | · | | · | | | |
| | | | Course Objective | es | | | |
| 1. | To demonstrate the basic laboratory techniques and application oriented physical | | | | | | |
| | consta | | | | | | |
| 2. | To pre | pare organic dye | es, organic compounds a | nd home care products. | | | |
| 3. | | | ss of water, DO, availab | le chlorine in bleaching | powder and | | |
| | saponi | fication value of | an oil. | | | | |
| | | | | | | | |
| | | | Course Outcomes (| CO) | | | |
| | CO1 | Gain the know | ledge of physical consta | ants and preparation of d | yes. | | |
| | CO2 | Know about th | ne preparation of organic | compounds. | | | |
| K1 – K5 | CO3 | Learn about th | e preparation method of | home care products. | | | |
| 111 113 | CO4 | Learn about es | stimation of hardness of | water, dissolved oxygen | ., | | |
| İ | | Saponification | of oil and isolation of c | itric acid. | | | |
| | CO5 | Learn about estimation of hardness of water, dissolved oxygen | | | | | |

| Prog | ramme C | Code: 04 | | B.Sc. Chemistry | | |
|---------|--|----------------|--|--------------------------|---------------------|--|
| Tit | le of the | paper | MAJOR ELECTIVE - I POLYMER TECHNOLOGY | | | |
| | Batch | ı | Hours / Week | Total Hours | Credits | |
| | 2023 - 20 | 024 | 4 | 60 | 5 | |
| | | | Course Objective | | | |
| 1. | To kno | w about basics | of polymers, polymerizat | tion and plastic materia | ıls | |
| 2. | 2. To learn about polymer processing and synthesis of some commercially important polymers and to know about various polymer processes techniques. | | | | | |
| 3. | To know different type of plastics, advancements, disposal, applications | | | | ons | |
| | | | Course Outcomes (Course Outcomes) | CO) | | |
| | CO1 | | the types of polymers, plications and uses. | , chemical and physic | cal properties, its | |
| V1 V5 | CO2 | | Understand the various polymerization techniques, processing and different types of individual polymer products. | | | |
| K1 – K5 | CO3 | Know about | different Polymerization F | Processing Techniques | | |
| | CO4 | Acquiring kar | nowledge of commercial | ly important polymer | products and its | |
| | CO5 | Know about | the recent advances in pol | ymer products and thei | r applications. | |

| Prog | ramme C | Code: 04 | B.Sc. Chemistry | | | | |
|---------|------------|------------------|---|---|------------------|--|--|
| Tit | tle of the | paper | MAJOR ELECTIVE - II NANO AND GREEN CHEMISTRY | | | | |
| | Batch | , | Hours / Week | Total Hours | Credits | | |
| | 2023 - 20 | | 4 | 60 | 3 | | |
| | 2023 - 20 | 024 | 4 | 00 | 3 | | |
| | | | Course Objective | s | | | |
| 1. | To gair | n knowledge al | out in - depth look at the | basics of Nano Chemis | stry and to know | | |
| 1. | the me | thods to prepar | e Nano materials. | | | | |
| 2. | To get | the knowledge | about Green Chemistry a | nd its limitations. | | | |
| 3. | To hav | e a holistic ide | a about Green solvents in | en solvents in laboratory as well as in Industry and also | | | |
| J. | to stud | y the Reactions | s and applications of Green | n Chemistry. | | | |
| | | | | | | | |
| | | | Course Outcomes (C | CO) | | | |
| | CO1 | To understan | d the basics of Nano Chen | nistry. | | | |
| | CO2 | To know the | methods to prepare Nano | materials. | | | |
| K1 - K5 | CO3 | To have an ic | lea about Nano chemistry | in medicine. | | | |
| | CO4 | To gain know | vledge about Green reaction | ons in laboratory. | | | |
| | CO5 | To gain know | vledge about Green solven | ts. | | | |

| Prog | ramme (| Code: 04 | | | | B.Sc. Ch | emistry | | | |
|---------------|---|---|--------------|--|-------|------------|----------|-----|------------|----|
| Tit | le of the | paper | | MAJOR ELECTIVE - III PHARMACEUTICAL CHEMISTRY | | | | | | |
| | Batch | 1 | Hour | s / Week | • | Tota | al Hours | | Credits | |
| | 2023 - 2 | 024 | | 4 | | | 60 | | 5 | |
| | Course Objectives | | | | | | | | | |
| 1. | 1. To know about the common diseases and cure-terms of pharmacology and drug action | | | | n. | | | | | |
| 2. | To get | get introduced to chemotherapy – antibiotics. | | | | | | | | |
| 3. | To kno | ow the drugs m | eant for dia | betes. | | | | | | |
| | | | Cours | e Outcor | mes (| (CO) | | | | |
| | CO1 | Gain the keepharmacolog | _ | about | the | common | diseases | and | cure-terms | of |
| K1 – K5 | CO2 | Understandir | ng Mechani | sm of act | ions | of drugs | | | | |
| $ X_1 - X_2 $ | CO3 | Understand a | about drug o | classificat | tion. | | | | | |
| | CO4 | Learn about | Common be | ody ailme | ents. | | | | | |
| | CO5 | Basic ideas a | bout variou | ıs health j | prom | oting drug | S. | | | |

| Prog | Programme Code: 04 | | B.Sc. Chemistry | | | |
|---------|--|---|--|---------------------------|---------|--|
| Tit | le of the | naner | MAJOR ELECTIVE - IV | | | |
| 110 | ic of the | paper | AGRICULTURAL CHEMISTRY | | | |
| | Batch | 1 | Hours / Week | Total Hours | Credits | |
| | 2023 - 20 | 024 | 4 | 60 | 5 | |
| | | | | | | |
| | | | Course Objective | S | | |
| 1. | To know about origin, physical and chemical aspects of soil | | | | | |
| 2. | To kno | To know about the basic idea of plant nutrients | | | | |
| 3. | To acquire the knowledge pesticides, fungicides and Herbicides | | | | | |
| | | | | | | |
| | | | Course Outcomes (Course Outcomes) | C O) | | |
| | CO1 | To gain the k | nowledge about the origin | ı soil. | | |
| | CO2 | To understan | d about physical and chen | nical properties of soil. | | |
| K1 – K5 | CO3 | To understan | d about chemical aspects | of soil | | |
| | CO4 | To learn abou | it plant nutrients. | | | |
| | CO5 | To learn Bas | o learn Basic ideas about pesticides, fungicides and herbicides. | | | |

| Prog | ramme C | Code: 04 | B.Sc. Chemistry | | | | |
|---------|--|--|--------------------------------------|--------------------------|--------------------|--|--|
| Tit | Title of the paper | | MAJOR ELECTIVE -V DAIRY CHEMISTRY | | | | |
| | Batch | 1 | Hours / Week | Total Hours | Credits | | |
| , | 2023 - 20 | 024 | 4 | 60 | 5 | | |
| | Course Objectives | | | | | | |
| 1. | To kno | w the chemistr | y of milk and milk produc | ets | | | |
| 2. | To kno | To know the basics of milk proteins, milk lipids, milk carbohydrates, and milk vitamins. | | | and milk vitamins. | | |
| 3. | To acquire knowledge of dairy products, analyze the constituents of milk products. | | | | nilk products. | | |
| | | | Course Outcomes (C | CO) | | | |
| | CO1 | Learning the | chemistry of milk and m | ilk products | | | |
| | CO2 | Knowing the basics of milk proteins, milk lipids, milk carbohydrates, and milk vitamins. | | | hydrates, and | | |
| K1 – K5 | CO3 | Understandi | ng the production and con | nposition of milk produ | ucts. | | |
| | CO4 | ' ' ' ' ' | the acquired knowledge of | of dairy products, analy | ze the | | |
| | | | of milk products. | | | | |
| | CO5 | To know co | mmercial values of milk. | | | | |

| Prog | ramme (| Code: 04 | B.Sc. Chemistry | | | | |
|---------|---|----------------------------------|---|-------------------------|----------------------|--|--|
| Tit | Title of the paper | | MAJOR ELECTIVE - VI LEATHER CHEMISTRY | | | | |
| | Batch | 1 | Hours / Week | Total Hours | Credits | | |
| | 2023 - 2 | 024 | 4 | 60 | 5 | | |
| | Course Objectives | | | | | | |
| 1. | To obta | ain the knowle | dge on the structure and co | omposition of the hides | s, skin and leather. | | |
| 2. | To know the basic principles involved in the pre-training methods of leather manufacture. | | | | | | |
| 3. | To und | lerstand about | vegetable tanning, chrome | tanning and leather ma | achinery. | | |
| | | | Course Outcomes (C | CO) | | | |
| | CO1 | Learning the | basic principles involved | in the theory of curing | hides and skins | | |
| | CO2 | Understandin | g the basics of soaking an | d bating process | | | |
| K1 – K5 | CO3 | Widening a s | Widening a skill on the bating and liming | | | | |
| | CO4 | Gaining the l | proad idea on the Chrome | tanning and vegetable | tanning process. | | |
| | CO5 | To know about leather machinery. | | | | | |

Subject Code: 23UCH5X1

| | | | | Subject Co | de: 23UCH5XI | |
|--------------------|------------|---------------------------------|--|----------------------------|---------------------|--|
| Programme Code: 04 | | Code: 04 | B.Sc. Chemistry | | | |
| m: | TT-1 0.1 | | EXTRA DEPARTMENTAL COURSE (EDC) - | | | |
| 111 | tle of the | paper | CHEMIST | RY IN DAY TODAY | LIFE | |
| Batch | ı | Semester | Hours / Week | Total Hours | Credits | |
| 2023 - 2 | 024 | V | 2 | 30 | 3 | |
| | • | | | | | |
| | | | Course Objective | es | | |
| 1. | To gai | n knowledge ab | out water treatment in inc | dustrial plant and its usa | age. | |
| 2. | To get | the knowledge | about industrial fermenta | ation process, oil, wax a | nd soap | |
| preparation. | | ation. | | | | |
| 3. | To hav | e a holistic idea | about food adulteration, | food hygiene and paint | ts manufacture. | |
| | | | | | | |
| | | | Course Outcomes (| C O) | | |
| | CO1 | Basic understa | derstanding of water technology and acquire knowledge in | | | |
| | COI | the treatment | of water for multi-purpos | se. | | |
| | CO2 | To study about Vitamins in Food | | | | |
| K1 – K5 | CO3 | To understand | the chemistry involved | in the manufacturing pr | ocess of oil, fats, | |
| K1 - K3 | COS | wax and soap. | | | | |
| | CO4 | To design a de | emonstration, that provid | es an opportunity to ide | entify adulteration | |
| | 004 | in food standa | rds. | | | |
| | CO5 | Broadening th | Broadening the knowledge about paints and pigments. | | | |

Subject code: 23UHR3N1

| | | | | Subject co | ode: 23UHR3N1 | |
|-----------|--------------------|---|--------------------------------|--|-------------------|--|
| Prog | Programme Code: 04 | | B.Sc. Chemistry | | | |
| T:4 | Title of the paper | | PART IV -NON MAJOR ELECTIVE -I | | | |
| 110 | ie oi uie | e paper | Н | UMAN RIGHTS | | |
| Batch | ı | Semester | Hours / Week | Total Hours | Credits | |
| 2023 - 20 | 024 | III | 2 | 30 | 2 | |
| | | | | | | |
| | | | Course Objective | S | | |
| 1. | To pre | epare for respon | sible citizenship with awa | reness of the relationsh | nip between | |
| 1. | Huma | n Rights, demo | cracy and development. | | | |
| 2. | To im | part education of | on national and internation | al regime on Human R | Lights. | |
| 3. | To ser | nsitive students | to human suffering and pr | omotion of human life | with dignity. | |
| 4. | To de | To develop skills on human rights advocacy. | | | | |
| 5. | To app | preciate the rela | tionship between rights ar | nd duties. | | |
| 6. | To fos | ster respect for t | colerance and compassion | for all living creature. | | |
| | | | | | | |
| | | | Course Outcomes (C | CO) | | |
| | CO1 | To understan | d the hidden truth of Hum | an Rights by studying | various theories. | |
| | CO2 | To acquire or | verall knowledge regarding | g Human Rights given | by United Nation | |
| | | Commission | (UNO) | | | |
| | CO3 | To gain know | vledge about various orgar | ns responsible for Hum | an Rights such as | |
| K1 – K5 | | National Hur | nan Rights Commission a | nd state Human Right o | commission | |
| | | (UNHCR). | | | | |
| | CO4 | To get habits | of how to treat aged perso | on, others and positive | social | |
| | | responsibiliti | es. | | | |
| | CO5 | To treat and | confirm, child, refugees ar | , child, refugees and minorities with positive social justice. | | |

Subject code: 23UWR4N2

| • | | | | • | de: 23UWR4N2 |
|-------------------------------|--------------------|------------------|-----------------------------------|--------------------------|-------------------|
| Prog | Programme Code: 04 | | B.Sc. Chemistry | | |
| Tie | Title of the paper | | Part IV -NON- MAJOR ELECTIVE – II | | |
| 110 | | | W | OMEN'S RIGHTS | |
| Batch | ı | Semester | Hours / Week | Total Hours | Credits |
| 2023 - 2 | 024 | IV | 2 | 30 | 2 |
| | | | | | |
| | | | Course Objective | S | |
| 1. | To kn | ow about the la | ws enacted to protect Wor | nen against violence. | |
| 2. | To im | part awareness | about the hurdles faced by | Women. | |
| 3. | To de | velop a knowle | dge about the status of all | forms of Women to acc | cess to justice. |
| 4. | To cre | eate awareness a | about Women's rights. | | |
| 5. | To kn | ow about laws a | and norms pertaining to pr | otection of Women. | |
| 6. | To un | derstand the art | icles this enables the Won | nen's rights. | |
| 7. | To un | derstand the Sp | ecial Women Welfare law | S. | |
| 8. | To rea | alize how the vi | olence against Women put | s an undue burden on l | healthcare |
| 0. | servic | es. | | | |
| | | | | | |
| | | | Course Outcomes (C | <u> </u> | |
| | CO1 | Understand t | he importance of Women' | s Studies and incorpora | ate Women's |
| | | Studies with | other fields. | | |
| | CO2 | Analyze the | realities of Women Empov | werment, Portrayal of V | Women in Media, |
| | | Developmen | t and Communication. | | |
| K1 – K5 | CO3 | Interpret the | laws pertaining to violence | e against Women and l | egal |
| $\mathbf{K}_1 - \mathbf{K}_2$ | | consequence | S. | | |
| | CO4 | Study the im | portant elements in the Inc | lian Constitution, India | n Laws for |
| | | Protection of | Women. | | |
| | CO5 | To be Aware | of Government Developm | nental schemes for wor | nen and to create |
| | | Awareness o | n modernization and impa | ct of technology on Wo | omen. |

| Prog | Programme Code: 04 | | B.Sc. Chemistry | | | |
|-------------------------------|--------------------|--------------------|---|----------------------------|--------------------|--|
| Tit | Title of the paper | | NON- MAJOR ELECTIVE – | | | |
| | | F F | CON | NSUMER AFFAIRS | | |
| | Batch | 1 | Hours / Week | Total Hours | Credits | |
| | 2023 - 2 | 024 | 2 | 30 | 2 | |
| | | | | | | |
| | | | Course Objective | | | |
| 1. | To fam | niliarize the stud | lents with their rights and | l responsibilities as a co | nsumer. | |
| 2. | To und | lerstand the pro | cedure of redress of cons | umer complaints. | | |
| 3. | To kno | ow more about o | decisions on Leading Cas | es by Consumer Protect | tion Act. | |
| 4. | To get | more knowledg | ge about Organizational s | et-up under the Consum | ner Protection Act | |
| 5. | To imp | oart awareness a | about the Role of Industr | y Regulators in Consun | ner Protection | |
| 6. | To und | lerstand Conten | nporary Issues in Consum | ner Affairs | | |
| | | | | | | |
| | | | Course Outcomes (| C O) | | |
| | CO1 | Able to know | the rights and responsibi | lity of consumers. | | |
| | CO2 | Understand t | he importance and benefi | ts of Consumer Protecti | ion Act. | |
| K1 – K5 | CO3 | Applying the | role of different agencies | in establishing product | and service | |
| $\mathbf{K}_1 - \mathbf{K}_2$ | | standards. | | | | |
| | CO4 | Analyse to ha | ndle the business firms' i | interface with consumer | ·s. | |
| | CO5 | Assess Qualit | ssess Quality and Standardization of consumer affairs | | | |

| Prog | Programme Code: 04 | | B.Sc. Chemistry | | | | |
|---------|---|---|--|------------------------|---------|--|--|
| Tit | Title of the paper | | JOB ORIENTED COURSE (JOC) – TEXTILE CHEMISTRY | | | | |
| | Batch | 1 | Hours / Week | Total Hours | Credits | | |
| | 2023 - 20 | 024 | 3 | 45 | | | |
| | Course Objectives | | | | | | |
| 1. | 1. To know about manufacture and properties of natural fibres (vegetable fibres, anim fibres) and synthetic fibres. | | | e fibres, animal | | | |
| 2. | To lear | arn preparatory process before dying. | | | | | |
| 3. | To know the principles of bleaching and dyeing. | | | | | | |
| | | | Course Outcomes (C | CO) | | | |
| | CO1 | Gain the kno | wledge about both synthet | ic and natural fibres. | | | |
| | CO2 | To know abo | out Regenerated And Synth | netic Fibres. | | | |
| K1 – K5 | CO3 | Understand about scouring and desizing. | | | | | |
| | CO4 | Learn about | bleaching. | | | | |
| | CO5 | Basic ideas a | bout dyeing | | | | |

Subject code: 23UCH1A1/23UCH3A3

| | | | <u> </u> | ubject code: 250 Cm | IAII 23UCIISAS |
|-----------|--------------------|-------------------|--|------------------------|------------------|
| Prog | Programme Code: 04 | | B.Sc., Biotechnology (I Year), Physics (II year), Botany (II | | |
| 1105 | | | Year), | Biochemistry (II Year) |) |
| Т; | la of th | a nonar | AI | LLIED PAPER – I | |
| 110 | ie oi ui | e paper | C | CHEMISTRY - I | |
| Batch | 1 | Semester | Hours / Week | Total Hours | Credits |
| 2023 - 20 | 024 | I/ III | 4 | 60 | 4 |
| | | | | | |
| | | | Course Objective | S | |
| 1. | To un | derstand the fun | damentals of Chemical bo | onding. | |
| 2. | To stu | ıdy various type | ly various types of organic Reaction. | | |
| 3. | To stu | ıdy the basic pri | nciples of thermodynamic | s and electrochemistry | |
| | • | | | | |
| | | | Course Outcomes (C | CO) | |
| | CO1 | Understandin | g the fundamental aspects | of chemical bonding a | and Interhalogen |
| | COI | compounds. | | | |
| 17.1 17.5 | CO2 | To acquire kr | nowledge of types for orga | nnic reaction | |
| K1 – K5 | CO3 | Study on the | various concepts in Thern | nodynamics. | |
| | CO4 | Study on the | various concepts in Electr | ochemistry. | |
| | CO5 | Acquiring kn | owledge about Fuel gases | and Petroleum. | |

Subject code: 23UCH2A2/23UCH4A4

| Subject code: 25UCH2A2/ 25UCH4A4 | | | | | |
|----------------------------------|--------------------|--|---|-----------------------------|-------------|
| Prog | Programme Code: 04 | | = - | (I Year), Physics (II year) | |
| | | | Year), | Biochemistry (II Year) |) |
| Tr' | 1 6.1 | | AL | LIED PAPER – II | |
| 111 | tle of the | e paper | C | HEMISTRY - II | |
| Batch | 1 | Semester | Hours / Week | Total Hours | Credits |
| 2023 - 20 | 024 | II/ IV | 4 | 60 | 4 |
| | | | | | |
| | | | Course Objectives | S | |
| 1. | To kn | ow the fundame | entals of Coordination com | npounds. | |
| 2. | To lea | ırn about some r | n about some natural products, amino acids and proteins. | | |
| 3. | To stu | study about quantitative and qualitative analysis and synthetic polymer. | | | mer. |
| | | | | | |
| | | | Course Outcomes (C | CO) | |
| | CO1 | Understandin | g the fundamental aspects | and applications of co | ordination |
| | COI | Chemistry. | | | |
| | CO2 | | various heterocyclic comp | | |
| K1 - K5 | | acids which i | which include their classification, preparation and properties. | | |
| | CO3 | To gain know | ledge about amino acids a | and vitamins. | |
| | CO4 | To understan | d theoretical aspects of qu | antitative and qualitati | ve analysis |
| | CO5 | Acquire the k | nowledge about synthetic | polymers, fibers and p | olastics |

Subject code: 23UCH2AL/ 23UCH4AL

| Subject code: 250 CH2/H2/ 250 CH4/H | | | ILI ICC CII IIIL | | |
|-------------------------------------|--|---|--|------------------------|---------|
| Programme Code: 04 | | Code: 04 | B.Sc., Biotechnology (I Year), Physics (II year), Botany (II | | |
| Tiog | 1 logianime Code. 04 | | Year), | Biochemistry (II Year) | |
| Tit | le of the | nonor | ALLI | ED PRACTICAL – I | |
| 110 | ie oi tile | paper | VOLUMETRIC | C AND ORGANIC AN | NALYSIS |
| Batch | 1 | Semester | Hours / Week | Total Hours | Credits |
| 2023 - 20 | 024 | II/ IV | 3 | 90 | 4 |
| | • | | | | |
| | | | Course Objective | \mathbf{S} | |
| 1. | 1. To demonstrate the basic laboratory technique of titration. | | | | |
| 2. | To gain | n deep knowle | deep knowledge about analysis of organic substances. | | |
| 3. | To ide | ntify the functi | onal groups of unknown c | ompounds. | |
| | | | | | |
| | | | Course Outcomes (C | CO) | |
| | CO1 | Remember th | ne basics of volumetric titr | rations. | |
| | CO2 | Studying the | use of indicators for vario | us titrations. | |
| K1 – K5 | CO3 | Understandin | ng about preliminary analy | sis of organic compour | nds. |
| | CO4 | Identification | n of the functional groups. | | |
| | CO5 | Practice for getting accuracy in volumetric estimations | | | |

Subject code: 23CDM101

| | Disaster Management | | | | |
|---------|--|--|---------------------|---|--|
| T;+1 | Title of the paper: | | THEORY | 1 – DISASTER MANAGEMENT AND | |
| 110 | e or the | paper. | SU | STAINABLE DEVELOPMENT | |
| |] | Hours / Week | | Total Hours | |
| | | 4 | | 60 | |
| | | | | | |
| | | | Course Ob | jectives | |
| 1. | To understand the basic aspects of History and Case Studies of Disasters and | | | tory and Case Studies of Disasters and | |
| 1. | PipelineDisasters and oil Spills. | | | | |
| 2. | To learn about Climate Changes and Disasters and gain knowledge about Disaster | | | | |
| | Manag | gement Educa | tion. | | |
| 3. | To study about Concept and benefits of Corporate Social Responsibility (CSR). | | | | |
| | | | | | |
| | | | Course Outco | omes (CO) | |
| | CO1 | Understand | the History and Ca | ase Studies of Disasters | |
| | CO2 | | nd the Pipeline Di | sasters and oil Spills & Land degradation and | |
| IZ1 IZ5 | CO2 | Droughts. | | | |
| K1 – K5 | CO3 | Gain the kno | owledge about Cli | mate Changes and Disasters. | |
| | CO4 | Study the ba | sic principles of D | Disaster Management Education. | |
| | CO5 | Explore the Concept and benefits of Corporate Social Responsibility (CSR). | | | |

Subject code: 23CDM102

| Subject code. 25CDW102 | | | | | | |
|---|---|--|--|-------------------------------|--|--|
| | Disaster Management | | | | | |
| THEOR' | | | | 2 – DISASTER PREPAREDNESS AND | | |
| Title of the paper: | | | RESPONSE | | | |
| | Hours / Week | | | Total Hours | | |
| | 4 | | | 60 | | |
| | | | | | | |
| Course Objectives | | | | | | |
| 1. | To know about the region of Natural disasters and study Safety engineering and | | | | | |
| 1. | analysis techniques. | | | | | |
| 2. | To have insight about Natural disaster effects and fighting against threats and | | | | | |
| acquire knowledge about Health care and safety. | | | and safety. | | | |
| 3. | To know about National disaster relief strategy and general preparedness. | | | | | |
| | | | | | | |
| Course Outcomes (CO) | | | | | | |
| | CO1 | Understand the Types of disasters and causes of disasters. | | | | |
| | CO2 | Gain the knowledge about Safety engineering and analysis techniques. | | | | |
| K1 – K5 | CO3 | Study about the Natural disaster effects and fighting against threats. | | | | |
| K1 – K3 | CO4 | Know about Health care and safety. | | | | |
| | 1 (1)5 | | ne various National disaster relief strategy and general | | | |
| | 003 | preparedness. | | | | |

Subject code: 23CDM103

| | | Di | saster Management | | |
|----------------------|---|-----------------------|---|--|--|
| Title of the paper: | | | THEORY 3 – DISASTER RECOVERY | | |
| |] | Hours / Week | Total Hours | | |
| | 4 60 | | | | |
| Course Objectives | | | | | |
| 1. | To learn group Causes of disaster and study about Disaster recovery plan. | | | | |
| 2. | To know about Role of technology in disaster recovery management and study about Environmental disaster management. | | | | |
| 3. | To learn about Disaster management to psychological perspectives. | | | | |
| Course Outcomes (CO) | | | | | |
| | CO1 | Gain the knowledge a | bout Disaster recovery. | | |
| | CO2 | Understand the basic | aspects of Disaster recovery plan. | | |
| V1 V5 | CO3 | Analyze and apply Ro | le of technology in disaster recovery management. | | |
| K1 – K5 | CO4 | Understand about Brid | ef history of the environment movement. | | |
| | CO5 | To meet the contempor | orary challenges on Disaster management to tives. | | |

Subject code: 23CIM101

| Instrumental Methods Of Chemical Analysis | | | | | |
|---|---|--|-------------------------------|-------------|--|
| Title of the paper: THE | | THEOR | EORY 1 – ANALYTICAL CHEMISTRY | | |
| | Hours / Week | | | Total Hours | |
| | | 4 | | 60 | |
| Course Objectives | | | | | |
| 1. | To understand the key features of Analytical chemistry and know the basics of Errors, Accuracy and Precision. | | | | |
| 2. | To identify different Separation techniques and describe about important purification techniques. | | | | |
| 3. | To understand important of Analytical biochemistry and industrial process. | | | | |
| Course Outcomes (CO) | | | | | |
| | CO1 | | | | |
| K1 – K5 | CO2 | To know the Knowledge about basics of Errors, Accuracy and Precision. | | | |
| | CO3 | Analyze the importance of Separation techniques. | | | |
| | CO4 | Describe about various types of purification techniques. | | | |
| | CO5 | To gain knowledge about Analytical biochemistry and industrial process | | | |

Subject code: 23CIM102

| Subject code. 25CH/H02 | | | | | |
|---|--|---|-----------------------------|-------------|--|
| Instrumental methods of chemical analysis | | | | | |
| TD: d C d | | THE | THEORY 2 – SPECTROSCOPY AND | | |
| Title of the paper: | | | CHROMATOGRAPHIC TECHNIQUES | | |
| Hours / Week | | | | Total Hours | |
| 4 | | | | 60 | |
| · | | | | | |
| Course Objectives | | | | | |
| 1. | To stuc | tudy Ultra – Violet and visible spectroscopy and study about Infrared spectroscopy. | | | |
| 2. | To und | lerstand about Nuclear Magnetic Resonance (NMR) spectroscopy. | | | |
| 3. | To study about various types chromatography and understand experimental techniques | | | | |
| ٥. | of column chromatography, | | | | |
| | | | | | |
| Course Outcomes (CO) | | | | | |
| K1 – K5 | CO1 | Understanding the fundamental aspect Ultra – violet and visible spectroscopy. | | | |
| | CO2 | Learn about Fundamental concepts of Infrared spectroscopy | | | |
| | CO3 | Acquire the knowledge of Nuclear Magnetic Resonance (NMR) spectroscopy. | | | |
| | CO4 | Study on the various types of chromatography. | | | |
| | CO5 | To inculcate knowledge about Column chromatography. | | | |

Subject code: 23CIM1CL

| Subject code. 25 ChvireL | | | | | |
|---|---|--|--------------------------------|---------------------------------------|--|
| Instrumental methods of chemical analysis | | | | | |
| T:41 | TT' d C d | | | PRACTICAL - 1 INSTRUMENT AND CHEMICAL | |
| Title of the paper: | | | METHODS IN DAY TO DAY ACTIVITY | | |
| |] | Hours / Week | | Total Hours | |
| | 4 | | | 60 | |
| Course Objectives | | | | | |
| 1. | Transformation of theoretical knowledge gain to practical aspects and have experience | | | | |
| 2 | | n handling organic compounds. | | | |
| 2. | | To determine the dissolved oxygen in different types of water. | | | |
| 3. | To know about fat content in milk using Lactometer and analysis techniques and | | | | |
| | unders | understand about Estimation of Hardness of water. | | | |
| | | | | | |
| Course Outcomes (CO) | | | | | |
| | CO1 | Understand t Analysis. | he basic principles | of Instrumental Methods of Chemical | |
| 17.1 17.5 | CO2 | • | tric experiments. | | |
| K1 – K5 | CO3 | | of organic compounds. | | |
| | CO4 | Improve the accuracy of analysis. | | | |
| | CO5 | To gain knowledge about Biological Oxygen Demand (BOD). | | | |